

Peroxide-Forming Chemicals (PFCs)

Last Reviewed: February 2022

Peroxide-forming chemicals (PFCs) are some of the most potentially hazardous substances handled in laboratories. PFCs are most often flammable organic liquids which are capable of forming potentially explosive R-O-O-R' peroxide bonds (where R = organic group) upon exposure to air or oxidizing impurities. They can also form on the surfaces of certain alkali metals and their amides. Organic peroxides are sensitive to heat, light or friction and if allowed to accumulate to high enough concentrations can result in explosions. Some peroxide forming chemicals have inhibitors such as BHT to prevent peroxide formation, but even these inhibitors can be exhausted over time particularly if a bottle gets opened and closed repeatedly.



For these reasons, peroxide formers must be monitored very closely. The specifics for how to monitor and check these items vary depending on the class of peroxide former you have in your lab (see the Labeling and Storage section of this SOP). Peroxide levels must be checked regularly and a log of test results maintained.

If you find a peroxide former in your lab and cannot determine a received date and/or opening date or if you notice the formation of crystals, discoloration, cloudiness, film, or stratification of layers take the following actions immediately:

- Do not move or open the container.
- Alert others in lab as to the potential hazard.
- Post a sign warning others not to disturb the container.
- Contact the ESD Hazardous Materials Group at (706) 542-5801 for disposal assistance.

Personal Protective Equipment



Standard lab coats are required. Flame resistant lab coats should be considered when handling flammable liquids and other hazardous materials that are easily ignited.



Nitrile chloroprene gloves provide adequate typically protection against minor splashes. Consult with your PI or supervisor determine whether materials involved in your process require alternative hand protection.



Eye Protection

ANSI Z87.1 - Compliant safety glasses or safety goggles if a splash hazard is present.

Labeling & Storage

PFCs should be stored in a flammable storage cabinet with self-closing hinges or in a refrigerator rated for flammable storage. All PFCs must be stored away from oxidizers and should be marked with receiving date and opening date. If the receiving and opening date is not known, promptly dispose of as hazardous waste. They should be purchased in limited quantities with older materials used first. Additionally, they must be managed in accordance with the following guidelines:

<u>Class A:</u> Chemicals that form explosive levels of peroxides without concentration, even if unopened.

Storage and Management: Date upon receipt and upon opening. If unopened from manufacturer, keep up to 18 months or stamped expiration date, whichever comes first. After opening, it is recommended that these chemicals be discarded or evaluated for peroxides no more than 3 months after opening. If crystals are visible in the solvent or around the cap or (in the case of alkali metals and their amides), if you notice film or discoloration on the surface, follow the instructions on the first page of this SOP.

Organic

Divinyl ether Isopropyl ether Divinyl acetylene Vinylidene chloride Inorganic

Potassium metal Potassium amide Sodium amide (sodamide) Class B: Chemicals that are a peroxide hazard when concentrated through evaporation or distillation.

Storage and Management: Date upon receipt and upon opening. If unopened from the manufacturer, keep up to 18 months or stamped expiration date, whichever comes first. After opening, materials must be discarded or evaluated for peroxides within 12 months and every 6 months thereafter. If crystals are visible in the solvent or around the cap, follow the instructions on the first page of this SOP.

Acetal Dicyclopentadiene Isopropanol* Diethylene glycol dimethyl ether Methyl acetylene Cumene Cyclohexene Diethyl ether Methyl cyclopentane Cyclooctene Dioxane (p-dioxane) Methyl -I-butyl ketone Cyclopentene Ethylene glycol dimethyl ether Tetrahydrofuran Diacetylene Tetrahydronapthelene Furan

Vinyl ether

<u>Class C:</u> Chemicals that may autopolymerize as a result of peroxide accumulation.

Acrylic acid Methyl methacrylate Vinyl acetylene Acrylonitrile Vinyl chloride Styrene Butadiene Tetrafluoroethylene Vinyl pyridine Chloroprene Vinyl acetate Vinylidene chloride

Chlorotrifluoroethylene

Storage and Management: Date upon receipt and upon opening. If unopened from the manufacturer, keep up to 18 months or stamped expiration date, whichever comes first. After opening, materials with inhibitors must be discarded or evaluated for peroxides within 12 months; chemicals without inhibitors must be discarded as hazardous waste as soon as possible after the container is opened. If crystals are visible in the solvent or around the cap or (in the case of alkali metals and their amides), if you notice film or discoloration on the surface, follow the instructions on the first page of this SOP.

Peroxide Testing

If test strips are to be employed to determine peroxide levels, they should cover the range from 0 – 100 ppm. The following peroxide levels must be used to determine activities that are deemed safe.

0-25 ppm - Material is safe to use or distill.

26-99 ppm - Material is safe to use, but should not be distilled.

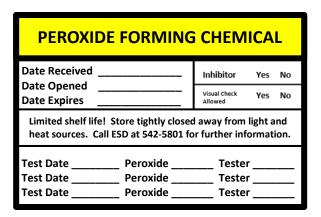
Above 100 ppm - Material should be disposed of and not used in lab.

^{*}These items will not form hazardous concentrations of peroxides unless they are concentrated. A visual check of these items must be performed and documented 12 months after opening and every 6 months thereafter. If the item will be used in distillation or evaporation then a peroxide test must be performed prior to performing the procedure and that test must be documented on the bottle.

A visual check may be performed for certain solid peroxide formers (for which testing strips are not appropriate) and some lower hazard liquids that only present a greater peroxide hazard upon distillation or concentration (see list above). A visual test must check for the following:

- Formation of crystals in the solution or around the cap
- Cloudiness or layer stratification within a liquid
- Discoloration or film on the surface of a solid

Any time a test (or visual check when allowed) is done on a substance, it must be documented. ORS provides the labels below and will affix these to bottles as they are found in the lab during inspections. These labels can also be requested by lab staff at any time.



Engineering Controls, Equipment & Materials

Fume Hood

Use a fume hood to keep exposure as low as possible when using these chemicals. A blast shield may also be considered in certain situations (e.g., distillations). If your protocol does not permit the handling of such materials ina fume hood, contact the Office of Research Safety (ORS) to determine whether additional respiratory protection is warranted.

Cautions & Considerations

Static Electricity

When transferring flammable liquids between containers greater than 4L (1 gallon) containers should be grounded, and the source container should be bonded to the receiving container during transfer. If possible, transfer flammable chemicals from glass containers to glassware or from glass container/glassware to plastic. Transferring these types of chemicals between plastic containers or unbonded metal containers may lead to a fire hazard due to static electricity.

	Housekeeping					
Spills	Please refer to Spill Control Guidelines for detailed information.					
Decontamination	Once any standing material has been wiped away, clean contaminated surfaces with soap and water. Dispose of contaminated paper towels as solid hazardous waste.					
	Any waste from this chemical class should be disposed of through the U Hazardous Waste Program. For assistance with arranging a waste pick you may contact the Environmental Safety Division (ESD) at 706-542-58 Prior to pickup, any container used to hold hazardous waste should labeled with the following:					
	- "Hazardous Waste"					
	- Chemical contents: Enough detail should be provided so that the full contents of the container are readily apparent. Labeling may include any of the following:					
Waste	 Percentages (Ex: 70% water, 30% hydrochloric acid) 					
	 Volumes (Ex: 1L of acetone, 500mL of water) 					
	Chemical classes (Ex: halogenated solvents)					
	Method (Ex: EPA 515.1 Herbicide Extraction Solvent Waste)					
	Referenced Log (Ex: See Laboratory Waste Log, Volume 2)					
	Utilizing Chematix waste profiles					
	 Any other labeling method providing enough detail to accomplish this requirement One or more of the following waste characteristics recognized by EPA: 					

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Ignitable, Corrosive, Reactive, or Toxic.

Use a Class BC or CO₂ extinguisher to put out a small fire.

Fire

Skin Eye Contact	Remove contaminated clothing and accessories; flush affected area with water. If symptoms persist, get medical attention.
Inhalation	Move person into fresh air. If symptoms persist, get medical attention.
Ingestion	Rinse mouth with water. If symptoms persist, get medical attention.

References

<u>Prudent Practices in the Laboratory: Handling and Management of Chemical Hazards,</u> National Research Council, 2011

"Management of Time Sensitive Chemicals: Their Identification, Chemistry, and Management," Chemical Health & Safety – American Chemical Society, November 2004, 17-24.

"Peroxides and Peroxide-Forming Compounds," Chemical Health & Safety – American Chemical Society, September 2001, 12-22.

"Review of Safety Guidelines for Peroxidizable Organic Chemicals," Chemical Health & Safety – American Chemical Society, 1996, 4(5), 28-36.

University of California – Center for Laboratory Safety

Contacts

Office of Research Safety: 706-542-5288 Environmental Safety Division: 706-542-5801